

Directions: Use the textbook, online textbook and others resources to help complete “In a Word,” “In a Sentence,” and “In a Picture.”

Vocabulary Practice

Study 5-10 minutes a Night

I.N. p.6

Text book L: Chemical Interactions

In a Word	In a Sentence	In a Picture	Word
Physical, mental, and other representations of an idea to help people understand a concept that they cannot observe. (p.8) _____			model
The smallest particle of an of any molecule. (p.7) _____			atom
Two or more atoms bonded together. (p.31) _____			molecule
A substance made of two or more different types of atoms. (p.7) _____			compound
A substance made up of one type of atom. (p.6) _____			elements
Very few atoms bonded together. _____			simple molecule

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I.N. p.7

Text book L: Chemical Interactions

In a Word	In a Sentence	In a Picture	Word
The force that holds atoms together. (p.13) _____			bonds
A substance that causes a chemical reaction. (p.57) _____			reactants
A substance formed due to a chemical reaction. (p.57) _____			products
The principle stating that matter is not created nor destroyed during a chemical reaction. (p.58) _____			law of conservation of mass
A reaction that absorbs energy in the form of heat (p.52) _____			endothermic reaction
A reaction that releases energy in the form of heat. (p.53) _____			exothermic reaction